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## Three new glycosides from *Hylocereus undatus*

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Three new glycosides, undatusides A–C (**1–3**), and 11 known compounds (**4–14**) were isolated from the flowers of *Hylocereus undatus*. Their structures were elucidated on the basis of spectroscopic data and chemical method.

**Keywords:** *Hylocereus undatus*; Cactaceae; undatusides A–C

### 1. Introduction

The plant *Hylocereus undatus* (Cactaceae) is widely distributed in the tropical areas of Africa, south America, and south Asia. The fruits of *Hylocereus* species, known as pitaya or dragon fruit, can be consumed freshly or processed as ingredient in juice, jams, or ice cream. In the southern China, the flowers of *H. undatus* have been used as food and folk medicine for the treatment of hyperactivity, cough, tuberculosis, mumps, bronchitis, and cervical lymph node tuberculosis for a long time. Previous phytochemical studies on this plant had resulted in the isolation of two triterpenes with microvascular protective activity [1]. As a part of the program to assess the chemical and biological diversity of the medicinal plants in the southern China, we carried out the chemical investigation of the flowers of *H. undatus*, which led to the isolation of three new glycosides, undatusides A–C (**1–3**), as well as 11 known compounds

(**4–14**) (Figure 1). Herein, we report the isolation and structural elucidation of the three new compounds.

### 2. Results and discussion

The molecular formula of **1** was established as C<sub>19</sub>H<sub>26</sub>O<sub>10</sub> by the pseudo-molecular ion in its HR-ESI-MS at *m/z* 437.1423 [M + Na]<sup>+</sup>. The UV spectrum of **1** showed the absorption maxima at 208 and 251 nm. The IR spectrum implied the presence of hydroxyl (3325 cm<sup>-1</sup>), carbonyl (1723 cm<sup>-1</sup>), and aromatic ring (1637 and 1455 cm<sup>-1</sup>). The <sup>1</sup>H NMR spectrum of **1** indicated the presence of five aromatic protons [ $\delta_{\text{H}}$  7.39 – 7.25 (5H, m)], three methylenes [ $\delta_{\text{H}}$  4.85 (1H, d, *J* = 12.0 Hz), 4.62 (1H, d, *J* = 12.0 Hz), 2.77 (1H, d, *J* = 15.6 Hz), 2.72 (1H, d, *J* = 15.6 Hz), and 2.67 (2H, s)], and a methyl [ $\delta_{\text{H}}$  1.38 (3H, s)]. Furthermore, the <sup>1</sup>H NMR spectrum displayed the signals due to an anomeric proton [ $\delta_{\text{H}}$  4.36 (1H, d, *J* = 7.5 Hz)] and a

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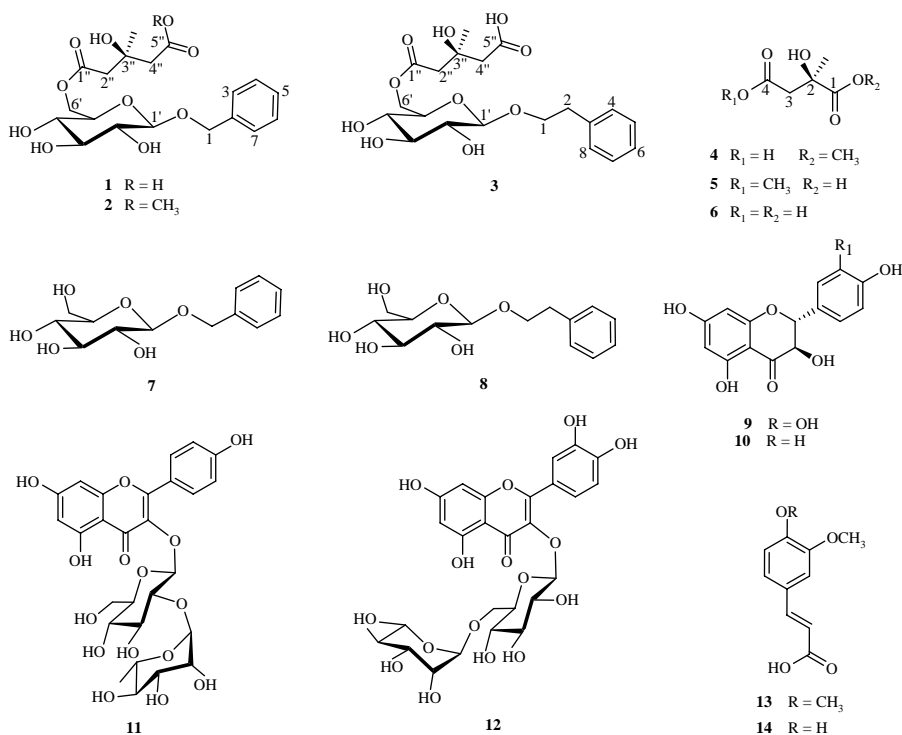


Figure 1. Chemical structures of compounds **1**–**14**.

hydroxymethyl group [ $\delta_{\text{H}}$  4.48 (1H, dd,  $J = 11.8, 2.0$  Hz) and 4.23 (1H, dd,  $J = 11.8, 5.7$  Hz)], as well as proton signals in the region  $\delta_{\text{H}}$  3.46 – 3.24 (4H, overlapped), suggesting the presence of a sugar moiety. Acid hydrolysis of **1** afforded D-glucose, which was identified by gas chromatography (GC) analysis using an authentic sample as reference. The  $\beta$ -configuration of D-glucose was determined based on the  $^3J_{\text{H}_1, \text{H}_2}$  coupling constant ( $J = 7.5$  Hz) of the anomeric proton. The  $^{13}\text{C}$  NMR and DEPT spectra of **1** revealed the presence of 19 carbon signals, including two carbonyl groups, a monosubstituted benzene ring, as well as a  $\beta$ -D-glucopyranosyl unit. With the aid of  $^1\text{H}$ – $^1\text{H}$  COSY, HSQC, and HMBC experiments, all the  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals of **1** were assigned as shown in Table 1.

The  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals of **1** assigned to **1a** were very similar to those of 3-hydroxy-3-methylglutaryl (HMG) moiety

[2], indicating that **1** possessed the same substructure, which was further supported by the HMBC correlations between H<sub>2</sub>-2'' [ $\delta_{\text{H}}$  2.77 (1H, d,  $J = 15.6$  Hz) and 2.72 (1H, d,  $J = 15.6$  Hz)] and C-4'' ( $\delta_{\text{C}}$  45.9)/3''-CH<sub>3</sub> ( $\delta_{\text{C}}$  27.7), between H-4'' [ $\delta_{\text{H}}$  2.67 (2H, s)] and C-2'' ( $\delta_{\text{C}}$  46.3)/3''-CH<sub>3</sub> ( $\delta_{\text{C}}$  27.7), as well as between 3''-CH<sub>3</sub> [ $\delta_{\text{H}}$  1.38 (3H, s)] and C-2'' ( $\delta_{\text{C}}$  46.3)/C-4'' ( $\delta_{\text{C}}$  45.9) (Figure 2). *S*-configuration for the C-3'' of the HMG moiety was assumed because naturally occurring HMG esters are formed via the acylation of the hydroxyl group with (*S*)-HMG-CoA [3,4]. Comparison of the NMR spectral data assigned to **1b** with those of benzyl- $\beta$ -D-glucopyranoside (**7**) suggested that they were very similar, except for the obvious downfield shift of C-6' and upfield shift of C-5', suggesting that the HMG moiety (**1a**) was esterified with the hydroxyl group at C-6' of the glucose moiety. Moreover, the HMBC correlations between H-6' ( $\delta_{\text{H}}$  4.48 and 4.23) and C-1'' ( $\delta_{\text{C}}$  172.4)

Table 1.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of compounds **1–3** ( $\text{CD}_3\text{OD}$ ,  $J$  in Hz)<sup>a,b</sup>.

Position	<b>1</b>		<b>2</b>		<b>3</b>	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$
1	71.8	4.85 (d, 12.0) 4.62 (d, 12.0)	71.8	4.85 (d, 12.0) 4.62 (d, 12.0)	71.8	4.01 (dt, 9.6, 7.3) 3.74 (dt, 9.6, 7.3) 2.90 (2H, t, 7.3)
2	138.8		138.9		37.1	
3	129.1	7.39	129.1	7.39	139.9	
4	129.2	7.32	129.2	7.32	129.9	7.24
5	128.7	7.25	128.7	7.25	129.3	7.24
6	129.2	7.32	129.2	7.32	127.1	7.16
7	129.1	7.39	129.1	7.39	129.3	7.24
8					129.9	7.24
1'	103.2	4.36 (d, 7.5)	103.3	4.36 (d, 7.5)	104.3	4.30 (d, 7.5)
2'	74.9	3.24	75.0	3.25	74.9	3.19
3'	77.7	3.36	77.8	3.36	77.7	3.36
4'	71.5	3.34	71.6	3.34	71.5	3.28
5'	75.9	3.46	75.2	3.44	75.1	3.45
6'	64.6	4.48 (dd, 11.8, 2.0) 4.23 (dd, 11.8, 5.7)	64.6	4.48 (dd, 11.8, 2.0) 4.22 (dd, 11.8, 5.7)	64.6	4.44 (dd, 11.8, 1.6) 4.20 (dd, 11.8, 5.8)
1''	172.4		172.4		172.4	
2''	46.3	2.77 (d, 15.6) 2.72 (d, 15.6)	46.4	2.76 (d, 14.4) 2.70 (d, 14.4)	46.4	2.72 (d, 14.6) 2.67 (d, 14.6)
3''	70.7		70.8		70.7	
4''	45.9	2.67 (2H, s)	45.9	2.70 (2H, s)	45.9	2.63 (2H, s)
5''	172.4		173.1		172.4	
5''-CH <sub>3</sub>	27.7	1.38 (3H, s)	27.8	1.38 (3H, s)	27.7	1.38 (3H, s)
5''-COOCH <sub>3</sub>			51.9	3.63 (3H, s)		

Notes: <sup>a</sup>Assignments were established by interpretation of the  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC, and HMBC spectra.<sup>b</sup>Overlapped signals are reported without designating multiplicity.

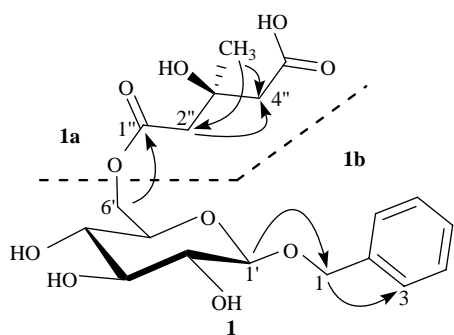


Figure 2. Key HMBC correlations of **1**.

confirmed the above deduction. Therefore, the structure of **1** was elucidated as benzyl-1-*O*-[6'-*O*-(*S*)-3''-hydroxy-3''-methylglutaryl]- $\beta$ -D-glucopyranoside and named as undatuside A.

The molecular formula of **2** was established as  $C_{20}H_{28}O_{10}$  by the pseudo-molecular ion in its HR-ESI-MS at  $m/z$  451.1584  $[M + Na]^+$ . The UV spectrum of **2** showed the absorption maxima at 211 and 258 nm. The IR spectrum implied the presence of hydroxyl ( $3377\text{ cm}^{-1}$ ), carbonyl ( $1729\text{ cm}^{-1}$ ), and aromatic ring ( $1687$  and  $1452\text{ cm}^{-1}$ ). Acid hydrolysis of **2** also afforded D-glucose. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of **2** were very similar to those of **1** except for the presence of methyl ester [ $\delta_{\text{H}}$  3.63 (3H, s);  $\delta_{\text{C}}$  173.1 and 51.9] instead of the aliphatic acid group. Thus, the structure of **2** was assigned as benzyl-1-*O*-[6'-*O*-(*S*)-3''-hydroxy-3''-methyl-5''-methoxyglutaryl]- $\beta$ -D-glucopyranoside and named as undatuside B.

The molecular formula of **3** was determined to be  $C_{20}H_{28}O_{10}$  by the pseudo-molecular ion in its HR-ESI-MS at  $m/z$  451.1584  $[M + Na]^+$ . Similar to **1** and **2**, the IR spectrum of **3** showed the presence of hydroxyl ( $3331\text{ cm}^{-1}$ ), carbonyl ( $1723\text{ cm}^{-1}$ ), and aromatic ring ( $1637$ ,  $1455\text{ cm}^{-1}$ ). Acid hydrolysis of **3** also afforded D-glucose. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data of **3** were closely similar to those of **1** except for the presence of an additional methylene [ $\delta_{\text{H}}$  2.90 (2H, d,

$J = 7.3\text{ Hz}$ );  $\delta_{\text{C}}$  37.1], indicating the presence of phenethyl unit in **3** instead of benzyl moiety. With the aid of  $^1\text{H}$ - $^1\text{H}$  COSY, HSQC, and HMBC experiments, all the  $^1\text{H}$  and  $^{13}\text{C}$  NMR signals of **3** were assigned as shown in Table 1. Furthermore, the HMBC correlations between H-1' ( $\delta_{\text{H}}$  4.30) and C-1 ( $\delta_{\text{C}}$  71.8), as well as between H<sub>2</sub>-6' ( $\delta_{\text{H}}$  4.44 and 4.20) and C-1'' ( $\delta_{\text{C}}$  172.4), suggested that the phenethyl and HMG units were connected to the C-1' and C-6' positions of glucose, respectively. Based on the above results, the structure of **3** was established as phenethyl-1-*O*-[6'-*O*-(*S*)-3''-hydroxy-3''-methylglutaryl]- $\beta$ -D-glucopyranoside and named as undatuside C.

In addition, the 11 known compounds were elucidated as (*R*)-(-)-citramalic acid 1-methyl ester (**4**) [5], (*R*)-(-)-citramalic acid 4-methyl ester (**5**) [6], (*R*)-(-)-citramalic acid (**6**) [7], benzyl- $\beta$ -D-glucopyranoside (**7**) [8,9], phenethyl- $\beta$ -D-glucopyranoside (**8**) [10], dihydroquercetin (**9**) [11], dihydrokaempferol (**10**) [12], kaempferol 3-neohesperidoside (**11**) [13], quercetin 3-*O*- $\beta$ -D-rutinoside (**12**) [14], *trans*-3,4-dimethoxycinnamic acid (**13**) [15], and *trans*-ferulic acid (**14**) [16], respectively, by comparison of their physical and spectroscopic data with those reported in the literature.

### 3. Experimental

#### 3.1 General experimental procedures

Optical rotation values were measured on a JASCO P-1020 digital polarimeter at room temperature. UV spectra were recorded on a JASCO V-550 UV/VIS spectrophotometer. IR spectra were measured on a JASCO FT/IR-480 plus Fourier transform infrared spectrometer with KBr pellets. 1D and 2D NMR spectra were recorded on Bruker AV-300 spectrometer. HR-ESI-MS data were acquired using an Agilent 6210 LC/MSD TOF mass spectrometer. ESI-MS data were determined on a Finnigan LCQ Advantage spectrometer. Column chromatographies (CC) were carried out

using silica gel (200–300 mesh, Qingdao Haiyang Chemical Group Corporation, Qingdao, China) and Sephadex LH-20 (Pharmacia Biotech AB, Uppsala, Sweden). TLC analysis was performed on precoated silica gel GF254 plates (Yantai Chemical Industrial Institute, Yantai, China). Analytical high-performance liquid chromatography (HPLC) was carried out on a Dionex chromatography equipped with a P680 pump, a PDA-100 photodiode array detector, and a Cosmosil 5C18-MS-II reversed-phase column (4.6 mm × 250 mm, 5.0 μm, Nacalai Tesque, Kyoto, Japan). Preparative HPLC was carried out on a Varian instrument equipped with a Prostar 215 pump, a Prostar 325 UV/VIS detector, and a Cosmosil 5C18-MS-II reversed-phase column (20 mm × 250 mm, 5.0 μm, Nacalai Tesque, Kyoto, Japan).

### 3.2 Plant material

The flowers of *H. undatus* were collected in Zhaoqing city, Guangdong Province of China, in February 2010, and authenticated by Prof. Guang-Xiong Zhou (Institute of Traditional Chinese Medicine & Natural Products, Jinan University). A voucher specimen (No. 20100215) has been deposited in the Institute of Traditional Chinese Medicine & Natural Products, Jinan University, Guangzhou, China.

### 3.3 Extraction and isolation

The air-dried flowers of *H. undatus* (15 kg) were powdered and extracted with 95% (V/V) EtOH at room temperature, and the solution was evaporated under vacuum to yield a residue (5700 g). The crude EtOH extract was subsequently suspended in distilled water and partitioned successively with petroleum ether, ethyl acetate, and *n*-butanol, respectively. After removing the solvent, the ethyl acetate extract (90 g) was subjected to silica gel CC using gradient mixtures of CHCl<sub>3</sub>–CH<sub>3</sub>OH

(100:0 → 0:100) as eluent to give four fractions (Fr. 1–4). Fr. 2 (30 g) was separated by silica gel column using CHCl<sub>3</sub>–acetone (100:0 → 0:100) as eluent to yield three subfractions (Fr. 2a–2c). Fr. 2a (14 g) was subjected to preparative HPLC [CH<sub>3</sub>OH–H<sub>2</sub>O (40: 60)] to give compounds **4** (188 mg), **5** (131 mg), and **6** (336 mg). Fr. 2b (11 g) was re-separated by Sephadex LH-20 column (CH<sub>3</sub>OH) and preparative HPLC [CH<sub>3</sub>OH–H<sub>2</sub>O (25: 75)] to yield compounds **1** (84 mg), **2** (23 mg), **3** (34 mg), **7** (1170 mg), **8** (129 mg), **9** (12 mg), **10** (14 mg), **11** (20 mg), **12** (21 mg), **13** (150 mg), and **14** (30 mg), respectively.

#### 3.3.1 Compound 1

Colorless oil:  $[\alpha]_D^{21} - 34.8$  (*c* 0.1, CH<sub>3</sub>OH); UV (CH<sub>3</sub>OH)  $\lambda_{\max}$  (log  $\epsilon$ ) 208 (3.88), 251 (2.58) nm; IR (KBr):  $\nu_{\max}$  3325, 2932, 2886, 1724, 1637, 1455, 1383, 1318, 1210, 1160, 1079, 1046, 1019, 905, 751, 700 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; HR-ESI-MS *m/z* 437.1423 [M + Na]<sup>+</sup> (calcd for C<sub>19</sub>H<sub>26</sub>O<sub>10</sub>Na, 437.1418).

#### 3.3.2 Compound 2

Colorless oil:  $[\alpha]_D^{21} - 29.8$  (*c* 0.2, CH<sub>3</sub>OH); UV (CH<sub>3</sub>OH)  $\lambda_{\max}$  (log  $\epsilon$ ) 211 (3.74), 258 (2.52) nm; IR (KBr):  $\nu_{\max}$  3377, 2944, 2874, 1729, 1687, 1452, 1354, 1323, 1209, 1158, 1078, 1049, 1019, 912, 742, 700 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; HR-ESI-MS *m/z* 451.1584 [M + Na]<sup>+</sup> (calcd for C<sub>20</sub>H<sub>28</sub>O<sub>10</sub>Na, 451.1575).

#### 3.3.3 Compound 3

Colorless oil:  $[\alpha]_D^{21} - 12.9$  (*c* 0.3, CH<sub>3</sub>OH); UV (CH<sub>3</sub>OH)  $\lambda_{\max}$  (log  $\epsilon$ ) 212 (3.59), 258 (2.38) nm; IR (KBr):  $\nu_{\max}$  3331, 2938, 2881, 1723, 1637, 1455, 1381, 1318, 1202, 1158, 1085, 1017, 908, 748, 700 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; HR-ESI-MS *m/z* 451.1584



$[M + Na]^+$  (calcd for  $C_{20}H_{28}O_{10}Na$ , 451.1575).

### 3.4 Acid hydrolysis and GC analysis of 1–3

Each solution of compounds 1–3 (each 1.5 mg) was heated in an ampoule with 1.5 ml of 2N HCl ( $CH_3OH-H_2O$ , 1:1) at 95°C for 2 h. The solution was evaporated with a stream of  $N_2$ . The reaction mixture was dissolved in  $H_2O$  and extracted with  $CHCl_3$ . The aqueous layer was evaporated to give a residue. Then, 2 ml of anhydrous pyridine and 3 mg of L-cysteine methyl ester hydrochloride were added to the residue, and the reaction mixture was heated at 60°C for 2 h. The solution was concentrated to dryness with  $N_2$ . Furthermore, *N*-(trimethylsilyl)imidazole (0.2 ml) was slowly added into the reaction mixture and then kept at 60°C for 1 h. Finally,  $H_2O$  (1 ml) was added to the solution to stop the reaction, and the aqueous layer was extracted with hexane (2 ml). The organic layer was analyzed using GC under the following conditions: column: HP-1701 (0.25 mm × 30 mm, 0.5  $\mu$ m), detector: FID, column temperature: 200–250°C (5°C/min), detector temperature: 280°C, injector temperature: 250°C, and carrier gas:  $N_2$ . The standard D-glucose and L-glucose were subjected to the same reaction and GC analysis under the above conditions [ $t_R$  (min): 32.231 (D-glucose), 34.863 (L-glucose)]. As a result, D-glucose [ $t_R$  (min): 32.185] was detected from the hydrolyzates of 1–3, respectively.

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